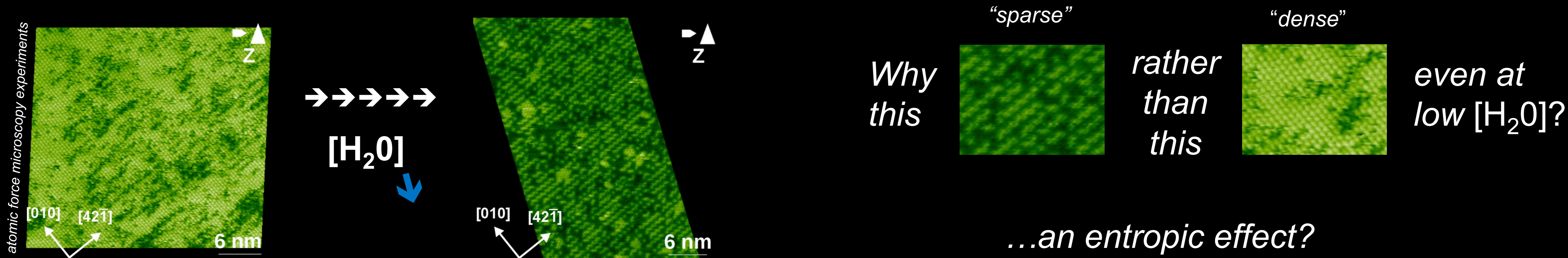
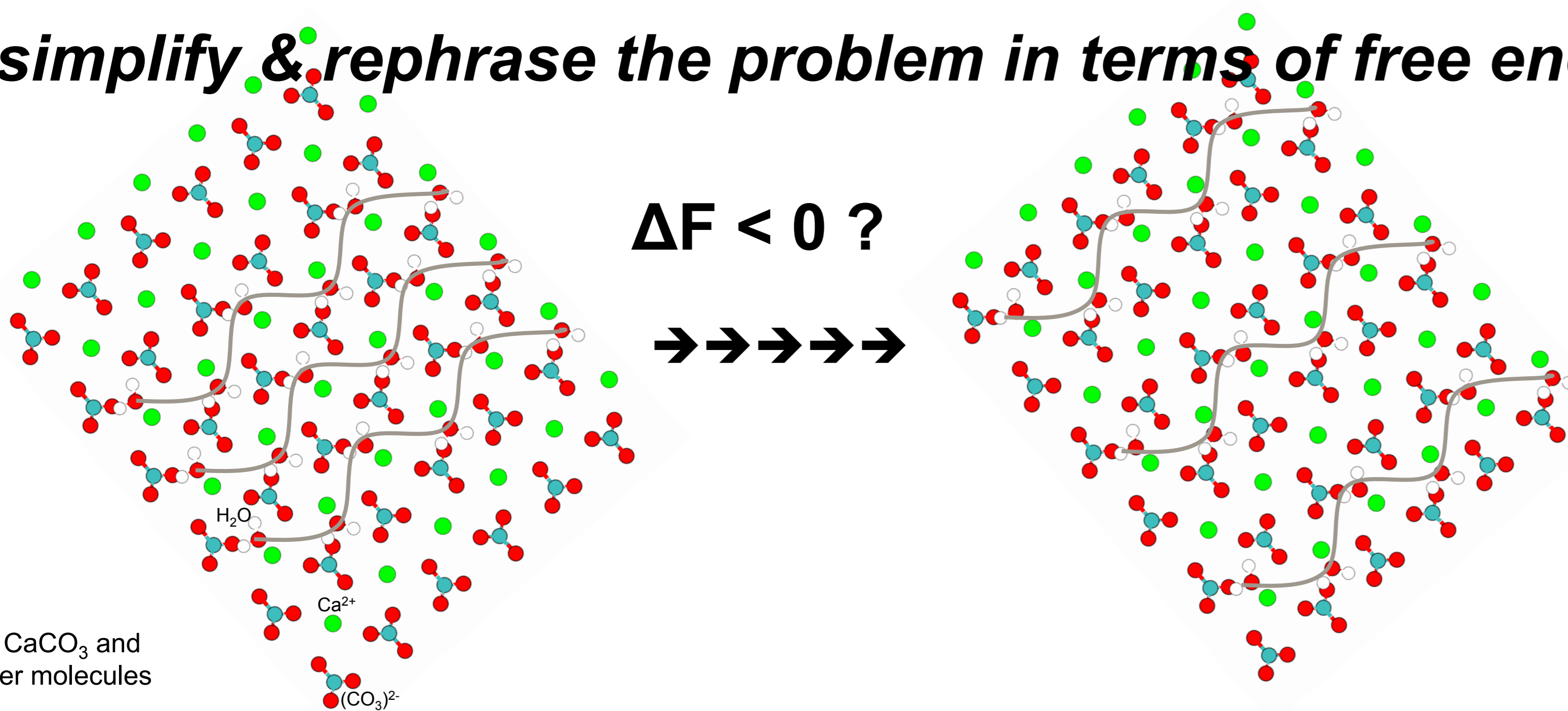


Entropy-driven self-assembly of water from first-principles?



simplify & rephrase the problem in terms of free energy



for a spontaneous transition, the free energy change ΔF is negative:

$$\Delta F = \Delta E - T\Delta S < 0$$

$$\rightarrow \Delta S > \Delta E/T$$

$$\Delta X = X(\text{sparse}) - X(\text{dense})$$

	ΔE	ΔS	$\Delta F = \Delta E - T\Delta S$
free energy truth table	negative	positive	always negative! best scenario!
	negative	negative	not entropically favored
	positive	positive	not energetically favored
	positive	negative	always positive! no transition!

in these cases $\Delta S > \Delta E/T$ determines what happens $\rightarrow \rightarrow \rightarrow \Delta E$ is easy to evaluate, ΔS very non-trivial!

T=0 K – energetics without entropy

$$\Delta E = +0.11 \text{ eV}$$

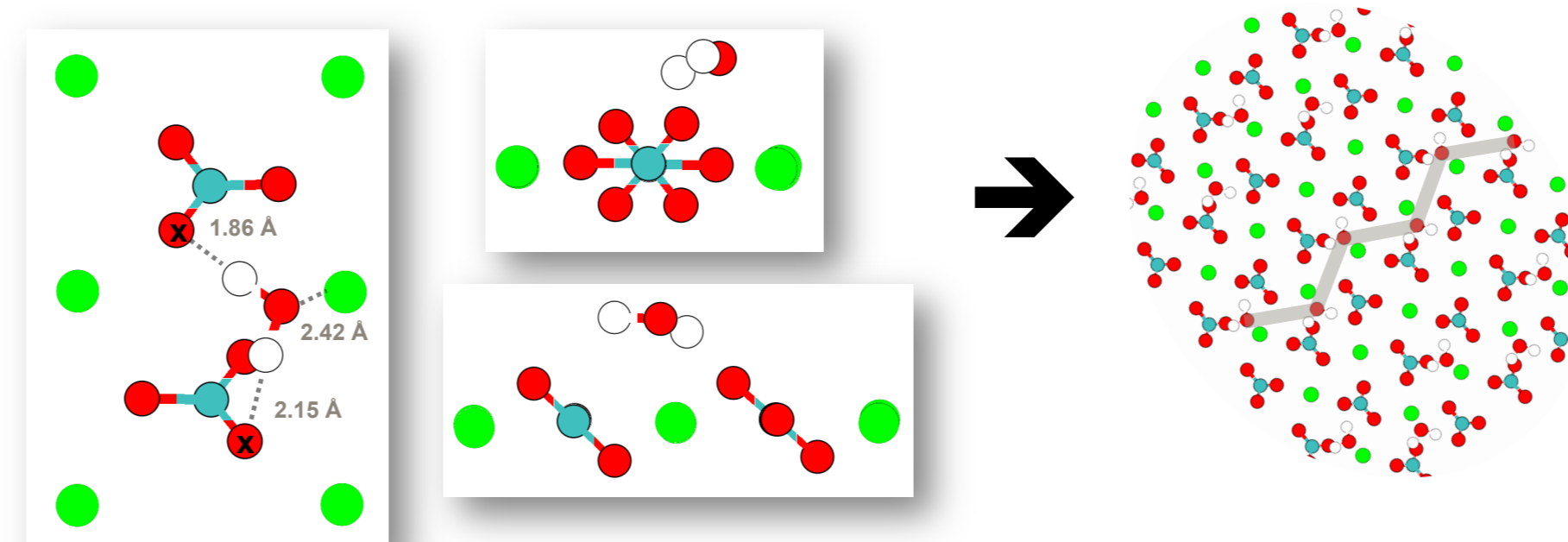
\rightarrow “dense” state is slightly favoured, but only slightly

adsorption energy $E_{\text{ads}} \sim 1 \text{ eV}$

\rightarrow agrees well with experimental extrapolation to T=0 K limit

bonding patterns

Ca^{2+} ions accommodate water oxygens and water hydrogens try to bond with $(\text{CO}_3)^{2-}$ oxygens \rightarrow zigzag pattern!



computational details

PBE-D3/DZVP/500 Ry, four layers of calcite + 18 waters = 720 atoms, 0.5 fs timestep in MD runs, 2 ps of equilibrium simulations at T = 300 K

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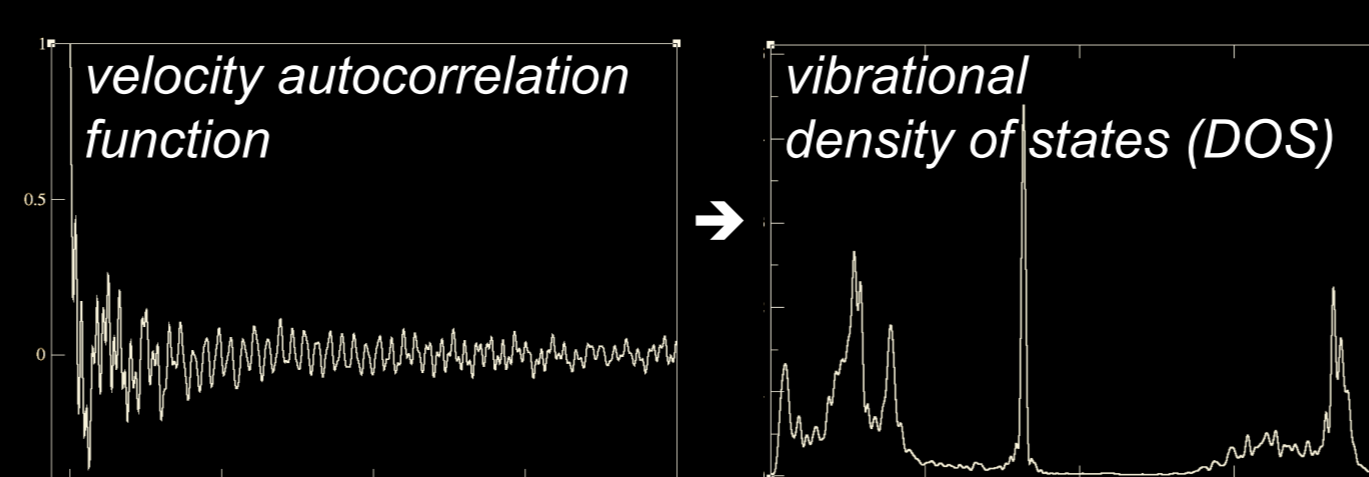
T=300 K – include thermal motion

$$\Delta E = -0.22 \text{ eV}$$

\rightarrow now the “sparse” state is slightly favoured, but again, only slightly

approximating entropy[†]

vibrational modes are independent quantum harmonic oscillators



$$\rightarrow \ln[Z] = \int \text{DOS}(\omega) \ln[z_{\text{qho}}] d\omega$$

$$\rightarrow S = k_B \ln[Z] + k_B T \partial_T \ln[Z]$$

$$\rightarrow \Delta S = +0.17 \text{ meV/K}$$

\rightarrow best case scenario! however

...very small differences, this yields $\Delta F \sim -0.3 \text{ eV}$...need longer trajectories

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