

# A?



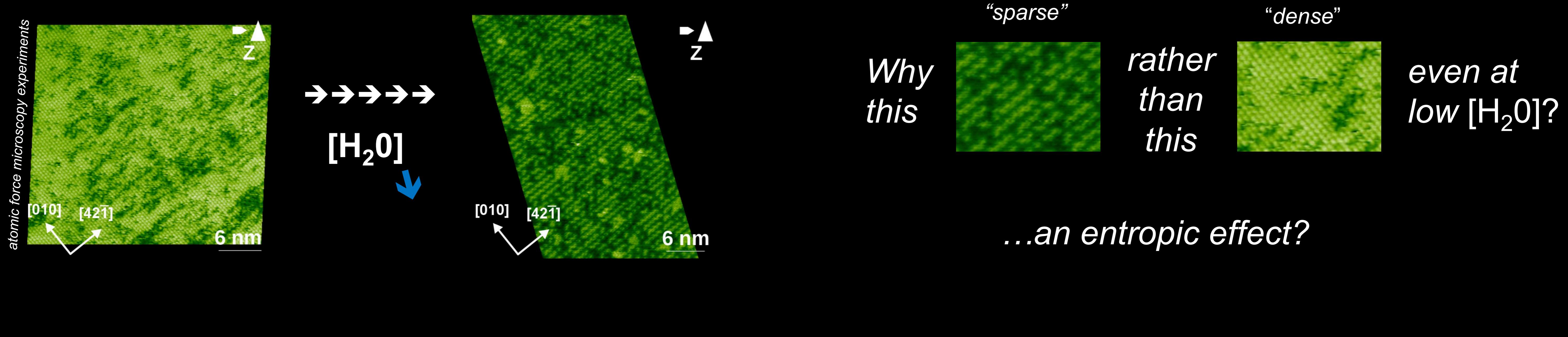
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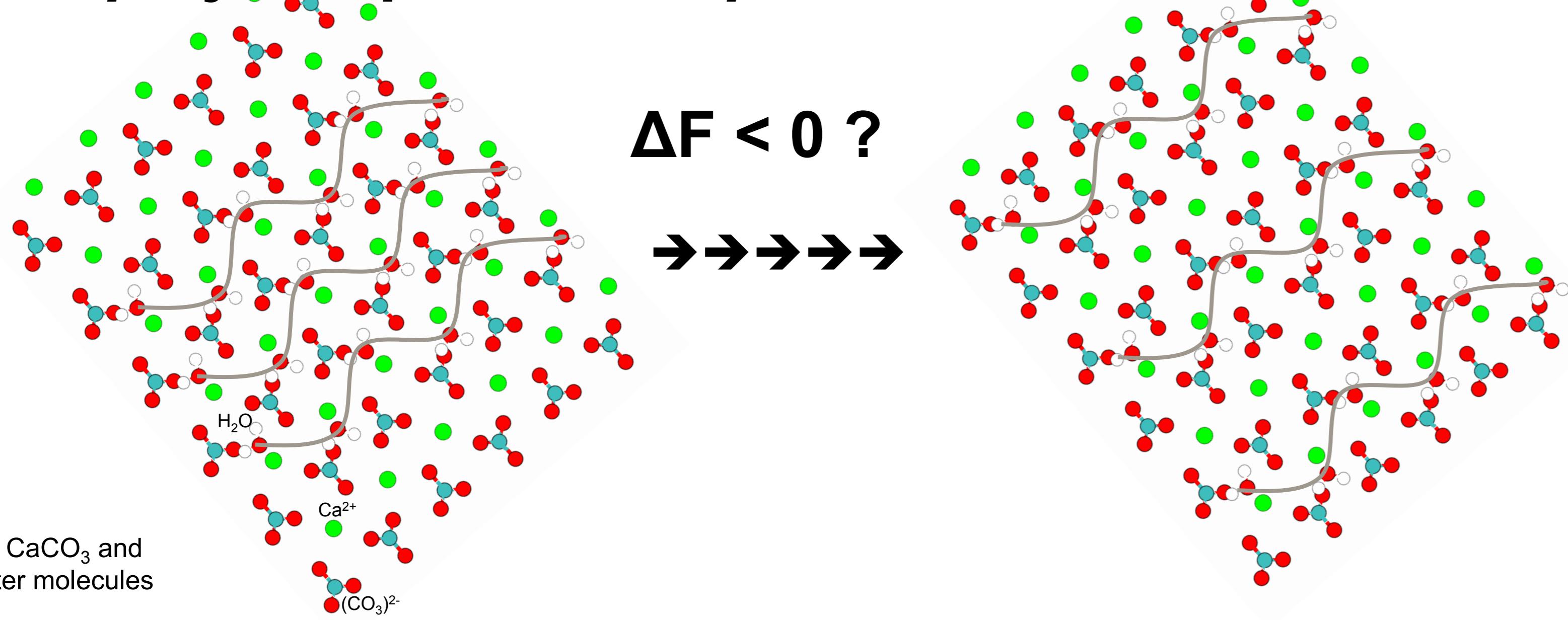
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# Entropy-driven self-assembly of water from first-principles?



simplify & rephrase the problem in terms of free energy



for a spontaneous transition,  
the free energy change  
 $\Delta F$  is negative:

$$\Delta F = \Delta E - T\Delta S < 0$$

$$\rightarrow \Delta S > \Delta E/T$$

$$\Delta X = X(\text{sparse}) - X(\text{dense})$$

	$\Delta E$	$\Delta S$	$\Delta F = \Delta E - T\Delta S$
free energy truth table	negative	positive	always negative! best scenario!
	negative	negative	not entropically favored
	positive	positive	not energetically favored
	positive	negative	always positive! no transition!

} in these cases  $\Delta S > \Delta E/T$  determines what happens  $\rightarrow \rightarrow \rightarrow$   $\Delta E$  is easy to evaluate,  $\Delta S$  very non-trivial!

## T=0 K – energetics without entropy

$$\Delta E = +0.11 \text{ eV}$$

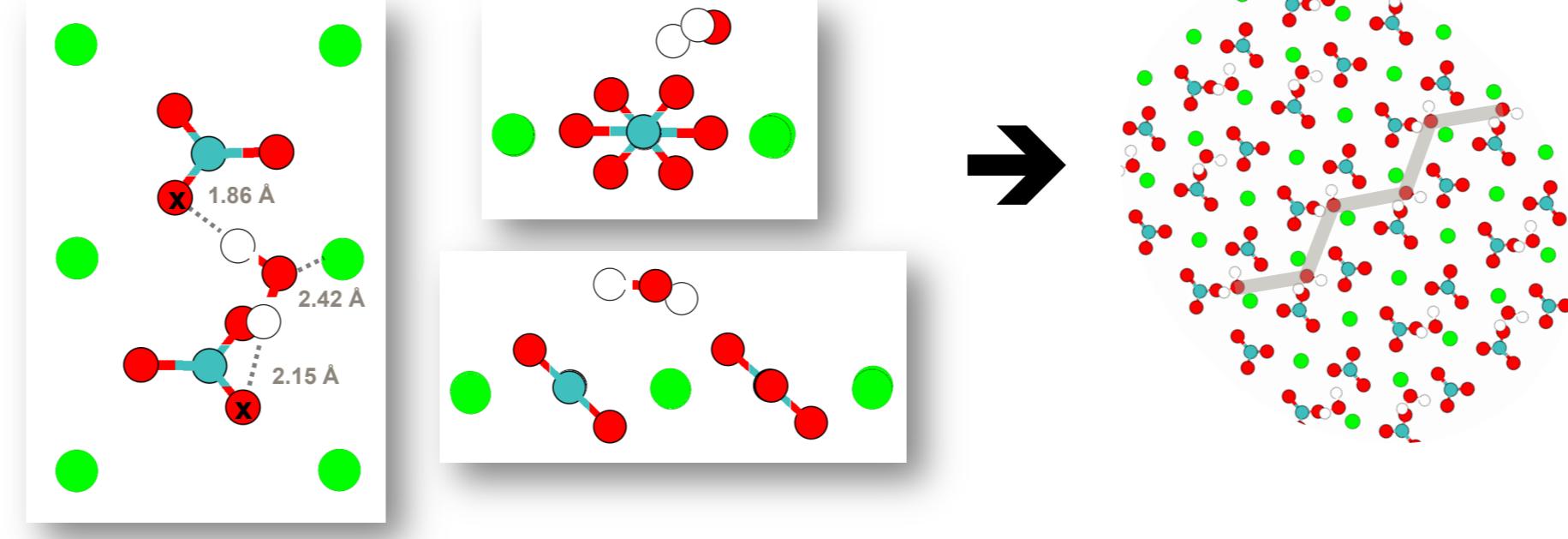
→ “dense” state is slightly favoured, but only slightly

adsorption energy  $E_{\text{ads}} \sim 1 \text{ eV}$

→ agrees well with experimental extrapolation to T=0 K limit

## bonding patterns

Ca<sup>2+</sup> ions accommodate water oxygens and water hydrogens try to bond with (CO<sub>3</sub>)<sup>2-</sup> oxygens → zigzag pattern!



## computational details

PBE-D3/DZVP/500 Ry, four layers of calcite + 18 waters = 720 atoms, 0.5 fs timestep in MD runs, 2 ps of equilibrium simulations at T = 300 K

[www.CP2K.org](http://www.CP2K.org)

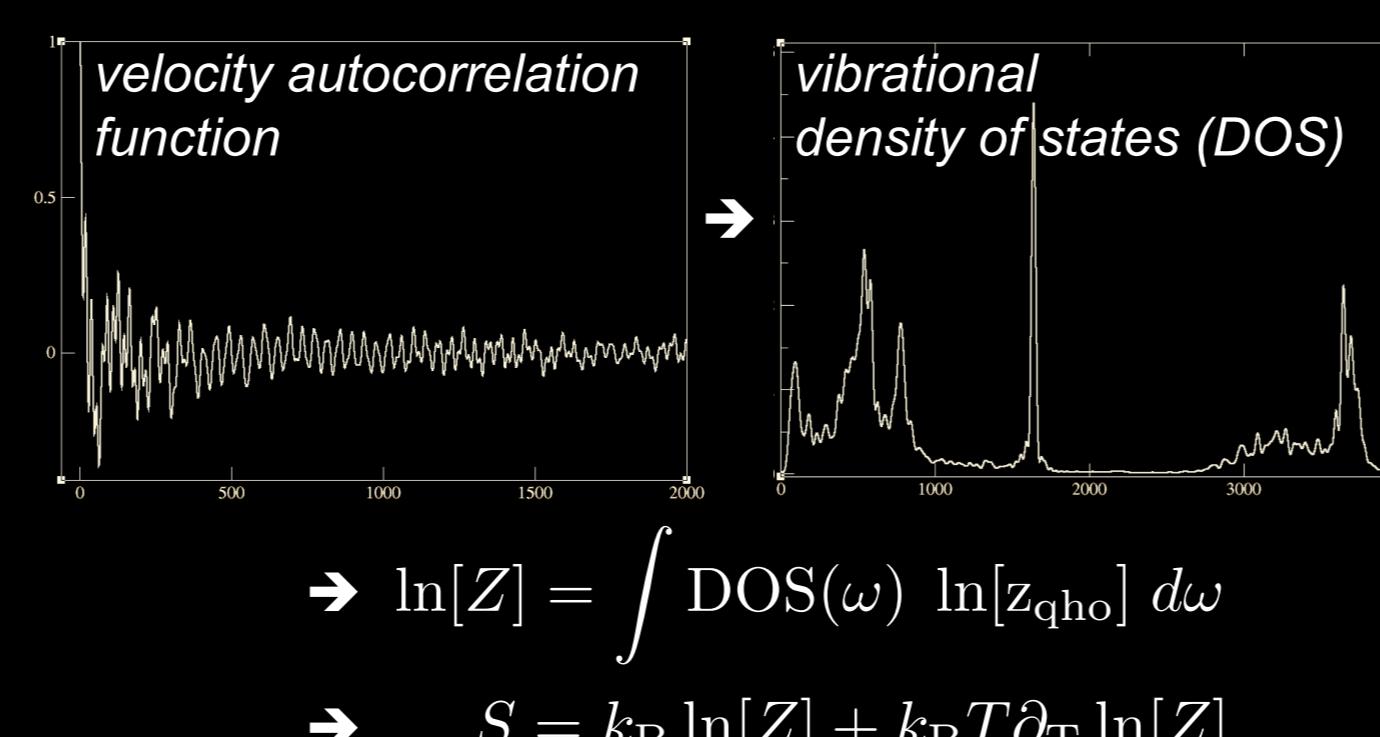
## T=300 K – include thermal motion

$$\Delta E = -0.22 \text{ eV}$$

→ now the “sparse” state is slightly favoured, but again, only slightly

## approximating entropy<sup>†</sup>

vibrational modes are independent quantum harmonic oscillators



$$\rightarrow \Delta S = +0.17 \text{ meV/K}$$

→ best case scenario!  
however  
...very small differences,  
this yields  $\Delta F \sim -0.3 \text{ eV}$   
...need longer trajectories

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